organic papers

(E)-1-[3-Ethoxy-4-(4-methylbenzenesulfonyloxy)-benzylidene]-2-(4-nitrophenyl)hydrazine acetonitrile solvate

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In the title compound, $C_{24}H_{24}N_4O_6S$, the ethylvanillin group makes dihedral angles of 59.15 (9) and 5.36 (8)° with the terminal 4-methylbenzene ring and the phenylhydrazine mean plane, respectively. The crystal structure is stabilized by intermolecular N—H···N hydrogen bonds and weak non-classical intermolecular C—H···O hydrogen-bonding interactions.

Comment

Metal complexes based on Schiff bases have attracted much attention because of their biological activity (Kahwa *et al.*, 1986). Consequently, a large number of Schiff base derivatives have been synthesized and employed to develop protein and enzyme mimics, such as models to mimic hydrolase in the hydrolysis of *p*-nitrophenyl picolinate (Li *et al.*, 2005). 1-(4-Nitrophenyl)hydrazine forms a variety of Schiff bases with aldehydes, and the synthesis and crystal structures of some of them, such as (E)-1-[4-(benzyloxy)benzylidene]-2-(4-nitro-phenyl)hydrazine (Jun, 2005) and (E)-1-(4-methoxy-3-propoxybenzylidene)-2-(4-nitrophenyl)hydrazine (Shi, 2005), have been reported.

(I)
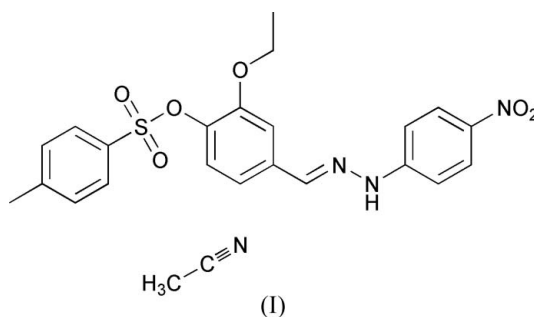
In the present study we report the synthesis and molecular structure of the nitrophenylhydrazine Schiff base derivative (I). In (I) (Fig. 1), all bond lengths and angles are within normal ranges (Allen *et al.*, 1987). The phenylhydrazine residue (C17–C22/N1–N3) is planar, with an r.m.s. deviation for fitted atoms of 0.0168 Å. This plane makes dihedral angles of 45.93 (9) and 11.68 (8)° with the C1–C6 benzene ring and the ethylvanillin group (C8–C13/C16/O1/O4), respectively. The dihedral angle between the C1–C6 benzene ring and the vanillin group is 40.35 (7)°. The nitro group (O5/N3/O6) and its attached aromatic ring are not coplanar, with a dihedral angle of 9.70 (4)°.

An intermolecular N—H···N hydrogen bond helps to consolidate the crystal packing (Table 1 and Fig. 2). There is also a weak C—H···O intramolecular hydrogen-bonding

interaction, linking the main molecule and the solvent molecule.

Experimental

An anhydrous ethanol solution (50 ml) of 2-ethoxy-4-formylphenyl 4-methylbenzenesulfonate (3.20 g, 10 mmol) was added to an anhydrous ethanol solution (50 ml) of 1-(4-nitrophenyl)hydrazine (1.53 g, 10 mmol) and the mixture was stirred at 350 K for 5 h under nitrogen, giving a yellow precipitate. The product was isolated and recrystallized from acetonitrile. Yellow single crystals of (I) suitable for X-ray analysis were obtained by slow evaporation of an acetonitrile solution.

Figure 1
The molecular structure of (I), with displacement ellipsoids for non-H atoms drawn at the 30% probability level.

Crystal data

$C_{22}H_{21}N_3O_6S \cdot C_2H_3N$	$V = 1237.6\ (5)$ Å^3
$M_r = 496.54$	$Z = 2$
Triclinic, $P\bar{1}$	$D_x = 1.332$ Mg m^{-3}
$a = 7.7257\ (17)$ Å	Mo $K\alpha$ radiation
$b = 7.8019\ (18)$ Å	$\mu = 0.18$ mm^{-1}
$c = 21.900\ (5)$ Å	$T = 294\ (2)$ K
$\alpha = 80.149\ (4)°$	Block, yellow
$\beta = 88.949\ (4)°$	$0.28 \times 0.26 \times 0.20$ mm
$\gamma = 72.208\ (3)°$	

Data collection

Bruker SMART APEX CCD area-detector diffractometer	6313 measured reflections
φ and ω scans	4332 independent reflections
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	3137 reflections with $I > 2\sigma(I)$
$T_{min} = 0.937$, $T_{max} = 0.965$	$R_{int} = 0.018$
	$\theta_{max} = 25.0°$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0677P)^2$
$R[F^2 > 2\sigma(F^2)] = 0.046$	$+ 0.202P]$
$wR(F^2) = 0.130$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 1.06$	$(\Delta/\sigma)_{max} = 0.003$
4332 reflections	$\Delta\rho_{max} = 0.19$ e Å^{-3}
319 parameters	$\Delta\rho_{min} = -0.32$ e Å^{-3}
H-atom parameters constrained	

Figure 2
Packing diagram for (I), with hydrogen bonds drawn as dashed lines.

Table 1
Hydrogen-bond geometry (Å, °).

$D—H\cdots A$	$D—H$	$H\cdots A$	$D\cdots A$	$D—H\cdots A$
$N2—H2\cdots N4^i$	0.86	2.25	3.094 (4)	167
$C24—H24A\cdots O2$	0.96	2.54	3.229 (3)	129

Symmetry code: (i) $x - 1, y, z$.

H atoms were included in calculated positions and refined using a riding-model approximation: C—H = 0.93 Å and $U_{iso}(H) = 1.2U_{eq}(C)$ for Csp^2; C—H = 0.97 Å and $U_{iso}(H) = 1.2U_{eq}(C)$ for methylene; C—H = 0.96 Å and $U_{iso}(H) = 1.5U_{eq}(C)$ for methyl; N—H = 0.86 Å and $U_{iso}(H) = 1.2U_{eq}(N)$ for amino H.

Data collection: SMART (Bruker, 1999); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997a); program(s) used to refine structur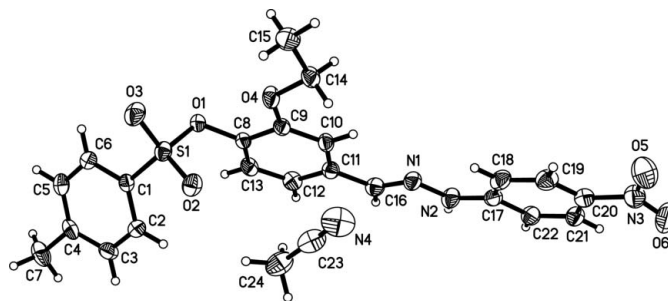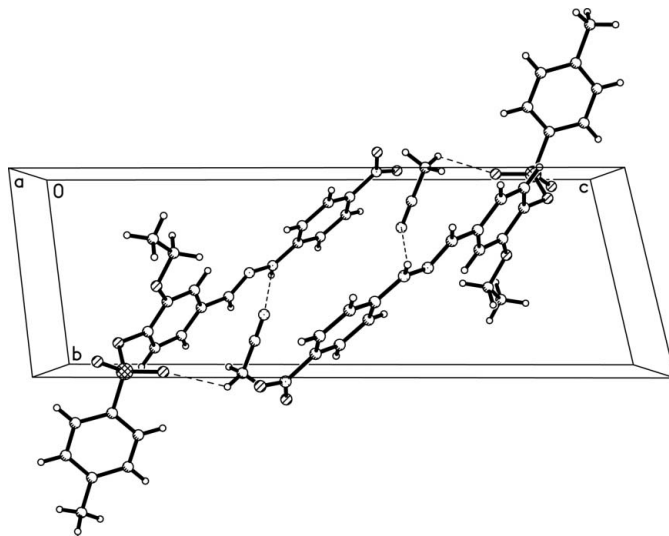e: SHELXL97 (Sheldrick, 1997a); molecular graphics: SHELXTL (Sheldrick, 1997b); software used to prepare material for publication: SHELXTL.

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